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1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At

The Formation Of Ammonia ( $\text{NH}_3$ ) From Nitrogen And Hydrogen (see Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The Equation:  $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$  (1)  $\Delta H^\circ = -92 \text{ kJ mol}^{-1}$   $\Delta S^\circ = -192 \text{ J mol}^{-1} \text{ K}^{-1}$

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