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3, Benzenesulfonic Acid Is Produced. $\text{SO}_3 \text{H}$ Benzenesulfonic Acid + SO_3 The Mechanism Of This Reaction Is Similar To That Of The Nitration Of Benzene. Concentrated $\text{H}_2 \text{SO}_4$ Is Used In An Initial Step To Generate The $\text{SO}_3 \text{H} +$ Electrophile As Shown. $\text{SO}_3 + \text{H}_2 \text{SO}_4 \rightarrow \text{SO}_3 \text{H} + \text{HSO}_4^-$ (i) Draw A Mechanism For The Reaction Of Benzene With $\text{SO}_3 \text{H}^+$... 3th, 2024

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