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3, Benzenesulfonic Acid Is Produced.  $\text{SO}_3\text{H}$  Benzenesulfonic Acid +  $\text{SO}_3$  The Mechanism Of This Reaction Is Similar To That Of The Nitration Of Benzene. Concentrated  $\text{H}_2\text{SO}_4$  Is Used In An Initial Step To Generate The  $\text{SO}_3\text{H}^+$  Electrophile As Shown.  $\text{SO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{SO}_3\text{H}^+ + \text{HSO}_4^-$  (i) Draw A Mechanism For The Reaction Of Benzene With  $\text{SO}_3\text{H}^+$  ... 3th, 2024

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Seen As The Starch Reacts With The Residual Iodine. Continue To Add FA 1 1 Cm<sup>3</sup> At A Time Until The Blue-black Colour Of The Starch-iodine Complex Disappears And 3th, 2024

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4 Alkanes And Alkenes Both React With Bromine. (a) Explain How And Why Bromine Can Be 1th, 2024

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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established.  $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$  What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The  $V^{2+}$  Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates 2th, 2024

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