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3, Benzenesulfonic Acid Is Produced. $\text{SO}_3 \text{H}$ Benzenesulfonic Acid + SO_3 The
Mechanism Of This Reaction Is Similar To That Of The Nitration Of Benzene.
Concentrated $\text{H}_2 \text{SO}_4$ Is Used In An Initial Step To Generate The $\text{SO}_3 \text{H}$ +
Electrophile As Shown. $\text{SO}_3 + \text{H}_2 \text{SO}_4 + \text{SO}_3 \text{H} + \text{HSO}_4^-$ (i) Draw A Mechanism
For The Reaction Of Benzene With $\text{SO}_3 \text{H}^+$... 3th, 2024

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Add Approximately 10 Drops Of Starch Indicator. A Blue-black Colour Should Be Seen As The Starch Reacts With The Residual Iodine. Continue To Add FA 1 1 cm^3 At A Time Until The Blue-black Colour Of The Starch-iodine Complex Disappears And
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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V^{2+} Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates 2th, 2024

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