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Equations Worksheet And Key 1. Answer The Following
Questions About The Chemical Equation Shown Below:
 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$ A) What Are The Reactants? B)
What Is The Product? C) What Do We Call The Number

"2" In Front Of The H₂ (and H₂O)? D) Is The Reaction
 Balanced? E) Why Is There Not A Coefficient For O₂? .
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 Following Equations: 1) $1 \text{ N}_2 + 3 \text{ F}_2 \rightarrow 2 \text{ NF}_3$ 2) $2 \text{ C}_6\text{H}_{10} + 17 \text{ O}_2 \rightarrow 12 \text{ CO}_2 + 10 \text{ H}_2\text{O}$ 3) $1 \text{ HBr} + 1 \text{ KHCO}_3$

1 H₂O + 1 KBr + 1 CO₂ 4) 2 GaBr₃ + 3 Na₂SO₃ 1
Ga₂(SO₃)₃ + 6 NaBr 5) 3 SnO + 2 NF₃ 3 SnF₂ + 1
N₂O₃ Using The Following Equation: 2 NaOH + H₂
SO₄ 2 H₂O + Na₂SO₄ 3th, 2024.

Flowchart For Balancing Equations Answer

KeyBalancing Equations Practice Worksheet Balancing
Of Chemical Equation In Telugu Chemical Reactions
And Equations Flowchart For Balancing Equations
Answer 3 Check Whether The Equation Is Balanced By
Counting The Atoms Of Each Element. 4 If Necessary,
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