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Agenda: 1) Go Over Quest 2) Introduce

Stoichiometry-S'more Activity HW: S'More Worksheet

#2 Tuesday, February 14

Agenda: Lab -Introduction To Stoichiometry

HW: 1) Lab Calculations 2) Read P

237-240 Wednesday, February 15

Agenda: 1) Go Over Lab 2) Worksheet #3

HW: 1) Read P 242 2) Video Notes Thursday, Jun 3th, 2024

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Practice Test Ch3 Stoichiometry (page 3 Of 3) 1. D It
Might Be Easiest To Balance The Equation With Mostly
Whole Numbers: $2 \text{NH}_3 + \frac{1}{2} \text{O}_2 \rightarrow 2 \text{NO} + 3 \text{H}_2\text{O}$.
The Question Asks About The Amount Of Oxygen
Reacting With ONE Mole Of Ammonia, Thus Cut The $\frac{1}{2}$
(3.5) Of Oxygen In Half To 1.75 2. C Balance: CH Jan
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