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Answer Key 1. A. 2 / 13 B. 13 / 8 C. 13 / 10 D. 2 / 8 (or 1 / 4) E. 2 / 10 (or 1 / 5) 2. The $\text{KClO}_3 / \text{O}_2$ Molar Ratio Is 2/3. 2 Mol $\text{KClO}_3 / 3$ Mol. $\text{O}_2 = 12.00$ Mol $\text{KClO}_3 / X = 18.00$ Mol. $X = 18.00$ Mol Of O_2 3. Given The Following Equation: $2 \text{K} + \text{Cl}_2 \rightarrow 2 \text{KCl}$ How Many Grams Of KCl Is Produced From 2.50 G Of K And ... 3th, 2024.

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3/7 [DOC] 1D Kinematics Review - With Answers #2
Answer: A. As An Object Falls, It Accelerates; This
Means That The Speed Will Be Changing. While Falling,
The Speed Increases By 10 M/s Every Second. The
Acceleration Is A Constant Value Of 10 M/s/s; Thus,
Choice B Should Not Be Chosen. 3th, 2024Chemistry
11 Stoichiometry Review Worksheet Name:

KEYChemistry 11 Stoichiometry Review Worksheet
Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actualyield=1.45g Theoyield=5.95g $\text{KClO}_4 \times 1\text{mol}$
 $122.55\text{g} \times 2\text{O}_2 \times 32.00\text{g} \times 1\text{mol} = 3.11\text{gO}_2$
%Yield= $1.45\text{g} / 3.11\text{g} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2$
 $\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percentpurity=62.8%?gpureFeCO₃ 4th, 2024.

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And Oxygen (O_2 .) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ C. 4th,
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