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Thermodynamics Example 9.2 The Element Mercury, Hg, Is A Silvery Liquid At Room  
Temperature. The Normal Freezing Point Of Mercury Is  $-38.9^{\circ}\text{C}$ , And Its Molar  
Enthalpy Of Fusion Is  $\Delta H$  3th, 2024 AP Chemistry Chapter 19 Chemical  
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Mercury Is A Silver Liquid At Room Temperature. Its Normal Freezing Point Is  
 $-38.9^{\circ}\text{C}$ , And Its Molar Enthalpy Of Fusion Is  $\Delta H_{\text{fusion}} = 2.29 \text{ kJ/mol}$ . W 1th, 2024.  
CHEM 1312. Chapter 17. Chemical Thermodynamics ...  $2\text{S}(\text{g})$ ,  $\Delta H^{\circ} = -20.2 \text{ kJ/mol}$   
And  $\Delta S^{\circ} = +43.1 \text{ J/K}\cdot\text{mol}$ . Which Of These Statements Is True? A. The Reaction Is  
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