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TEXT: All Classes Use The Same Textbook: Gilbert, Kirss, And Foster, Chemistry: An Atoms-Focused Approach, 1st Ed.; W. W. Norton: NY, 2014. CHEM 1091/1092 And CHEM 1101/1102: CHEM 1091 And 1101 (Fall Only) Topics Chapter In Text Properties Of Matter, Energy, Measurements, Uni 2th, 2024

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17.!Explain The Trend In Acidity. Include Key Structures. L The Stronger Acid Has A Lower Energy Conjugate Base. The Cyclopropene Is Least Acidic Because Its Conjugate Base Is Antiaromatic (high Energy, Loop Of P's, 4 π E-). The Linear Diene's Conjugate Base Is Resonance Stabilized, And Therefore

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