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### **PREPARATION AND TESTING OF BUFFER SOLUTIONS**

67 PREPARATION AND TESTING OF BUFFER SOLUTIONS. P. URPOSE. The Purpose Of The Laboratory Investigation Is To Experimentally Determine (1)  $pK_a$  (and Thus  $K_a$ ) Of The Acid In A Buffer And Thus The Buffer Range, (2) Investigate The

Buffer Capacity Of 1th, 2024

### **Section 19.1. Acid-Base Buffer Solutions**

If We Add  $\text{CH}_3\text{COO}^-$ -SHIFT Net Effect:  $[\text{H}^+]$  Decreases PH Increases Also,  $[\text{OH}^-]$  Increases POH Decreases I.e., a Weak Acid And Its Conjugate Base Or A Weak Base And Its Conjugate Acid. 19-2 Example: Addition Of  $\text{NH}_4\text{Cl}$  To  $\text{NH}_3$  1th, 2024

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