

## Chemical Calculations Stoichiometry Prentice Hall Answers Free Pdf

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 $C_1 \cdot V_1 = C_2 \cdot V_2$   
 $12 \cdot V_1 = 0.1 \cdot 1000$   
 $V_1 = 100/12 = 8.33$  Ml  
What Is Molarity Of Pure Water ?  
Molar Concentration: Moles Of Substance Per Liter Of Solution  
1 Liter Of Water Weighs 997 G At 25 °C  
FW Of H<sub>2</sub>O: 2+16=18  
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