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#### **Chemical Names And Formulas - Weebly**

The Name Of A Base Is The Name Of The Cation Followed By The Name Of The Anion (hydroxide). The Formula Of A Base Is Written By Showing The Number Of Hydroxide Ions Needed To Balance The Positive Charge On The Cation. After Reading Lesson 9.4, Answer The Following Questions. Names And Formulas Of Acids Hydrogen 1. 4th, 2024

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## **CHEMISTRY NOTES - Chapter 6 Chemical Names And Formulas**

14. Writing The Formulas Of Compounds. A. Binary And Ternary Ionic Compounds: Basically You Have To Work Backward From The Name To The Symbols. For Example Sodium Chloride Involves Sodium And Chlorine And The Formula Is NaCl. What You Need To Be Careful Of Is The Ratio Of Ions In The Compound. The Compound Must Be Neutral (have Equal Positive And 3th, 2024

#### Section 6.1 Chapter 6 Chemical Names And Formulas ...

1 Chapter 6 Chemical Names And Formulas Adapted From Notes By Stephen Cotton Section 6.1 Introduction To Chemical Bonding ZOBJECTIVES: -Distinguish Between Ionic And Molecular Compounds. -Define Cation And Anion And Relate Them To Metal And Nonmetal. Molecules And Mo 2th, 2024

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# **Chapter 6 Chemical Names And Formulas - Weebly**

56. Without Consulting Table 6.4, Name The Following Ions. 6.3 A. OH-hydroxide B. Pb4+ Lead(IV) C. SO 4 2-sulfate D. O2-oxide E. HPO 4 2-hydrogen Phosphate F. Cr 2 O 7 2-dichromate G. Al3+ Aluminum H. ClO 2-chlorite 62. Complete The Table Below By Writing Correct Formulas For The Compounds Formed By Combining Positive And Negative Ions. 4th, 2024

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