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4.  $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$  5.  $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$  Classification Of Chemical Reactions Reactions, Much Like Elements And Compounds, Can Be Classified According To Type. The Ability To Recognize And Classify Reactions Can Help Us Predict 1th, 2024

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The Reactants Are The Elements That React With One And Another In A Chemical Reaction, While The Products Are The Elements That We Get After The Reaction. The Chemical Equation Has The Products On The Right Side, While The Reactants Are Written On

The Left Side. Both Of Them Are Separated By An Arrow. 3th, 2024

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