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Hour10 When Vanadium(II) Compounds Are Dissolved In Water, The Following
Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The
Composition Of The Equilibrium Mixture In Favour Of The V^{2+} Ions? A Adding An
Acid B Adding A Reagent That Selectively Precipitates Feb 1th, 2024Chemistry 9701
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4443) Www.megalecture.com AS Chemistry: Atomic Structure Page 3 Question 2
9701_w/14/qp23 Jan 1th, 2024CHEMISTRY 9701/04(c) (i) Write A Chemical Equation
Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A

Value For The Lattice Energy Of AgBr(s). First Ionisation Energy Of Silver = +731kJmol⁻¹ Electron Affinity Of Bromine = 325kJmol⁻¹ Enthalpy Change Of Atomisation Of Silver = +285kJmol⁻¹ Enthalpy Change Of Atomisation Of Bromine = +112kJmol⁻¹ ... Mar 1th, 2024.

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