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Vapor Pressure Of The Liquid Is 0.013 Bar At 85.8–C And 0.053 Bar At 119.3–C, Use The Clausisu-Clapeyron Equation To Calculate (a) The Enthalpy Of Vaporization, (b) The Normal Boiling Point And (c) The Entropy Of Vapori 1th, 2024Solutions To Sample Quiz Problems And Assigned ProblemsFor A Monatomic Interacting Classical Gas, With Interactions That Only Depend On The Particle Co-ordinates, Derive The Maxwell Boltzmann Distribution Of Velocities And Show That The Average Kinetic Energy Is Given By = 3Nk BT=2. Solution. See Eqs. (94,95) Of The Notes. |||||{Quiz Problem 12. Using The Fact That E 2= 2= K BT2C V Show That E=Eis Proportional 1=N1=2. Solution. See Eqs. ... 1th, 2024.

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