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At 100°C H<sub>2</sub>O(g) At 140°C A-B B-C C-D D-E E-F. The Heating Curve Shown Above Is  
A Plot Of Temperature Vs Time. It Represents The Heating Of Substance X At A  
Constant Rate Of 2th, 2024 Lab #4- Heating Curve - Ms. Hart's Chemistry Class  
2013-2014 Versus Temperature Is Called The Heating Curve\*\*. Water Is A Common  
Substance. Ice Is The Stable Phase Below 0oC. Both Solids And Liquids Coexist At  
Exactly 0oC. What Does It Mean To Coexist? \_\_\_\_ When Heat Is Put Into The  
System, More Solid Will Melt. Thus, The Temperature Does Not Change at 0C  
During The Phase Change. The Normal Boiling ... 2th, 2024.  
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Substance, A Change In Temperature Or A Phase-change May Result. If Heat Is  
Added To A Substance At A Constant Rate, And The Temperature Is Plotted Against

Time, The Graph Produced Assumes 3th, 2024Heating Curve Of Water Lab  
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