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Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then,  $n(\text{N}) = 0.0322$ . aniline  $n(\text{Aceta II}) = 0.0322$ . We Find The Molar Mass Of Acetanilide:  $(\text{N}) = 128.19141161135$ . 89 M Aceta ... 2th, 2024

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