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A) $[\text{Co}(\text{H}_2\text{O})_6]^{2+} + \text{Co} + 2\text{e}^-$ E) $[\text{FeO}_4]^{2-} + \text{Fe} + 6\text{e}^-$ I) $\text{K}_2[\text{CoCl}_4] + \text{Co} + 2\text{e}^-$ B) $[\text{CrCl}_6]^{3-} + 3\text{e}^-$ F) $[\text{Mn}(\text{CN})_6]^{4-} + \text{Mn} + 2\text{e}^-$ J) $\text{K}_3[\text{AuF}_6] + \text{Au} + 3\text{e}^-$ C) $[\text{Co}(\text{NH}_3)_6]\text{Cl}_2 + \text{Co} + 2\text{e}^-$ G) $[\text{Ni}(\text{CO})_4] + \text{Ni} + 0\text{e}^-$ K) $(\text{NH}_4)_2[\text{IrCl}_6] + \text{Ir} + 4\text{e}^-$ D) $[\text{Co}(\text{NH}_3)_5\text{Cl}]\text{Cl}_2 + \text{Co} + 3\text{e}^-$ H) $[\text{Ni}(\text{EDTA})]^{2-} + \text{Ni} + 2\text{e}^-$ L) $\text{Na}[\text{Mn}(\text{CO})_5] + \text{Mn} + 1\text{e}^-$ TASK 2 - Writing Half Equations A) $2\text{I}^- \rightarrow \text{I}_2 + 2\text{e}^-$ B ... 2th, 2024

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Ethanol Has The Formula $\text{C}_2\text{H}_5\text{OH}$ And Is Used As A Fuel (e.g. For Cars In Brazil).
It Burns ... 4th, 2024

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Metal Carbonate \rightarrow Metal Oxide + Carbon Dioxide (on Heating) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations.
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