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A) [Co(H<sub>2</sub>O)<sub>6</sub>]<sup>2+</sup> + Co<sup>+2</sup> E) [FeO<sub>4</sub>]<sup>2-</sup> - Fe<sup>+6</sup> I) K<sub>2</sub>[CoCl<sub>4</sub>] Co<sup>+2</sup> B) [CrCl<sub>6</sub>]<sup>3-</sup> - Cr<sup>+3</sup> F) [Mn(CN)<sub>6</sub>]<sup>4-</sup> - Mn<sup>+2</sup> J) K<sub>3</sub>[AuF<sub>6</sub>] Au<sup>+3</sup> C) [Co(NH<sub>3</sub>)<sub>6</sub>]Cl<sub>2</sub> Co<sup>+2</sup> G) [Ni(CO)<sub>4</sub>] Ni<sup>0</sup> K) (NH<sub>4</sub>)<sub>2</sub>[IrCl<sub>6</sub>] Ir<sup>+4</sup> D) [Co(NH<sub>3</sub>)<sub>5</sub>]Cl<sub>2</sub> Co<sup>+3</sup> H) [Ni(EDTA)]<sup>2-</sup> - Ni<sup>+2</sup> L) Na[Mn(CO)<sub>5</sub>] Mn<sup>-1</sup> TASK 2 - Writing Half Equations A) 2I<sup>-</sup> → I<sub>2</sub> + 2e<sup>-</sup> B ... Jan 1th, 2024

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Metal Carbonate → Metal Oxide + Carbon Dioxide (on Heating) CaCO<sub>3</sub> → CaO + CO<sub>2</sub> TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. Apr 1th, 2024

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