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3) $2(aq) + C. K(s) + H_2SO_4(aq) \rightarrow$ Y D. $Zn(s) + NaOH(aq) \rightarrow$ Y SECTION 11.3 REACTIONS IN AQUEOUS SOLUTION 1. Write The Net Ionic Equation For The Reaction Between Aqueous Barium Nitrate, $Ba(NO_3)_2$, And Sodium Sulfate, Na_2SO_4 . 2. Magnesium Reacts With HCl To Form Hydrogen And Magnesium Chloride. Write The Balanced Net Ionic Equation For This Reaction. 3. 3th, 2024

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DESCRIBING CHEMICAL ...

A Chemical Reaction Can Be Concisely Represented By
A Chemical 1.. The Substances That Undergo A
Chemical Change Are The 2.. The New Substances
Formed In A Chemical Reaction Are The 3.. In
Accordance With The Law Of Conservation Of , 4. A
Chemical Equation Must Be Balanced. When Balancing

An 5. Equation, You Place In Front Of Reactants And ...
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6.1 - Describing Chemical Reactions

(b) This Is A Chemical Reaction As A Gas Is Created (carbon Dioxide) During The Reaction. (a) + Hacog (b) The Bottle Could Be Heated Up Or Opened To Reverse The Reaction. CHECK YOUR LEARNING Suggested Answers 1. The Arrow Shows Which Way The Chemical Reaction Is Going. 2. 2th, 2024

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Figure 11.1 The Separate Ingredi-ents Of The Bread

Are Now Combined Into One ProductÑthe Loaf Of
Bread. Connecting To Your World Section 11.1
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