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 Temperature We Will Once Again Look At The Formation Of Ammonia (NH_3) From
 Nitrogen And Hydrogen (see Section Chemical Equilibrium Of The Thermodynamics
 Chapter). This Reaction Follows The Equation: $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$ (1) $H_0 = 92 \text{ KJ Mol}$
 $S_0 = 192 \text{ J Mol K}$ 3th, 2024 Chemical Engineering, CM3510: Chemical Reaction
 ...Grading: Exam 1 15% Exam 2 15% Exam 3 15% Final Exam 20% Research Paper
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 Was Used: $A \geq 92$ $A/B \geq 89$ $B \geq 82$ $B/C \geq 80$ $C \geq 71$ $C/D \geq 69$ $D \geq 61$ $F \leq 60$ If
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