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$N\text{O} + n\text{e}^- \rightleftharpoons \text{NRR}$  (2.1.36) The Free Energy Of The Cell Reaction Is Given By,  $\Delta G = \Delta G_0 + RT \ln \frac{a_{\text{N}} a_{\text{H}^+} a_{\text{H}_2\text{O}}}{a_{\text{N}_2} a_{\text{H}_2}}$  (2.1.38) Which Since  $\Delta G = -nF E_{\text{rxn}}$  And  $a_{\text{H}^+} = a_{\text{H}_2} = 1$  Becomes,  $E = E_0 + \frac{RT}{nF} \ln \frac{a_{\text{N}}}{a_{\text{N}_2}}$  (2.1.40) This Expression Is Called The Nernst Equation. This Relation Can 3th, 2024

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This Gives Us The Magnitude, Or Size, Of The Impedance (in Ohms) Of This System.  
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The Carbon Electrode Surface By Using Na<sub>2</sub>SO<sub>4</sub> Electrolyte Solution (8). Ultrathin  
Metal Layers Deposited On The Foreign Metal Surfaces Are Well-known To Have  
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