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Distinguish Among Strong Electrolytes, Weak  
Electrolytes, And Nonelectrolytes, Giving Examples Of  
Each. Solvents And Solutes Solution - A Homogenous  
Mixture, That Is Mixed Molecule By Molecule. Solvent -  
The Dissolving Medium Solute-the Dissolved Particle  
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### **Chapter 1 Chemical Foundations - Department Of Chemistry**

6 A Conversion Factor • Is A Fraction Obtained From  
An Equality. Equality:  $1 \text{ in.} = 2.54 \text{ cm}$  • Is Written As A  
Ratio With A Numerator And Denominator. • Can Be  
Inverted To Give Two Conversion Factors For Every

Equality. 1 In. And 2.54 Cm 2.54 Cm 1 In. Conversion Factors Jul 1th, 2024

## **Name: Date: Foundations Of Chemistry Unit Practice Problems**

4. Write The Formula For The Following Covalent Compounds: A. Tetraphosphorus Triselenide F. Silicon Dioxide B. Disilicon Hexabromide G. Diboron Tetrabromide C. Diselenium Diiodide H. Xenon Hexafluoride D. Tetrasulfur Dinitride I. Dinitrogen Trioxide E. Hydrogen Gas J. Apr 2th, 2024

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How Elements Form Compounds Much Of Chemical Activity Involves The Transfer Of Electrons From One Substance To Another. Ionic Compounds Are Generally Combinations Of Metals And Nonmetals. These Compounds Are Held Together By Attractions Between Opposite Charged Ions (like A Magnetic Attraction). This Is Called An Ionic Bond. Jun 3th, 2024

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2. All Atoms Of A Given Element Are Identical\*, But The Atoms Of One Element Are Different From The Atoms Of All Other Elements. \*Since Disproved 3. Atoms Of One Element Cannot Be Changed Into Atoms Of A Different Element By Chemical Reactions; Atoms Are

Neither Created Nor Destroyed In Chemical Reactions.  
4. Jun 3th, 2024

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