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### **Worksheet 7 - Ideal Gas Law I. Ideal Gas Law Ideal Gas Law ...**

Worksheet 7 - Ideal Gas Law I. Ideal Gas Law The Findings Of 19th Century Chemists And Physicists, Among Them Avogadro, Gay-Lussac, Boyle And Charles, Are Summarized In The Ideal Gas Law:  $PV = nRT$   $P$  = Pressure  $V$  = Volume  $n$  = Moles Of Gas,  $R$  = Universal Gas Constant  $T$  = Temperature. The Value Of  $R$  Varies With The 3th, 2024

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A Perfect Radiator (“black Body”) Would Have  $E=1$ . A Perfect Reflector (“shiny” Object) Would Not Radiate At All;  $E=0$ . Lecture 33 10/28 Radiation This Behavior Is

Contained In The Stefan-Boltzmann Law: Here, E Is 1th, 2024

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And The SI Users Can Ignore Them. Problems With The Are Solved Using EES, And Complete Solutions Together With Parametric Studies Are Included On The Enclosed DVD. 3th, 2024

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Temperature, Pressure, And Volume. Which Gas Has A Greater Number Of Collision

Of Gas Molecules With The Walls Of The Container? 1. The He Gas Because It Is Less Massive And Moving With A Higher Average Velocity Correct 2. They Are The Same Since The Pressure Is The Same 3. The O2 Since It Has A Higher Average Momentum Since It Is More Massive 4. 4th, 2024

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