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2) Nylon Is Commonly Made By Step-growth Polymerization. In Class We Made Nylon 6,10. A) Give The Structure Of Nylon 6,10 Circling The Chemical Group That Defines This Polymer As Nylon. B) Draw The Structure Of The Two Chemicals That Were Used In Class To Make Nylon 6,10. C) Explain Why This Is A Condensation Polymerization And What Molecule ... 3th, 2024

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3.12. Polymers There Are Two Types Of Polymerisation: Addition And Condensation Addition Polymerisation An Addition Polymer Forms When Unsaturated Monomers React To Form A Polymer Monomers Contain C=C Bonds Chain Forms When Same Basic Unit Is Repeated Over And Over. Poly(a Lkenes) Are Chemically Inert Due To The Strong C-C And C-H 1th, 2024

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484 S. Roy Et Al. / Designed Monomers And Polymers 12 (2009) 483 495 Mers Can Be Broadly Classi Ed Into Two Groups, Namely Speci C And Non-speci C [15]. The Speci C Bioadhesive Polymers (e.g., Lectins, Mbrin) Have The Ability To Adhere To Speci C Chemical Structures Within The Biological Molecules While The 3th, 2024

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Polymers 2020, 12, 255 2 Of 18 Than  $L = D(\rho/D)^3$  [12,13]. For This Reason, The WLC Model Is Broadly Accepted As Kind Of A "gold Standard" As Far As The Statistical Mechanics Of Semiflexible Chains Is Concerned. Care Is Needed, However, When One Deals With The Adsorption Of Semiflexible Polymers On Planar Surfaces [14-20]. 1th, 2024

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