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 $(\partial V/\partial T)P = R/P$. Substitution $\Rightarrow \Delta S = C \ P \ Ln(T \ 2/T \ 1) - R \ Ln(P \ 2/P \ 1)$ Applying $\Delta S = 0$ $\Rightarrow Ln(T \ 2/T \ 1) = R/Cp \ Ln(P \ 2/P \ 1) \Rightarrow (T \ 2/T \ 1) = (P \ 2/P \ 1) (C \ P-C \ V)/C \ P \ T \ T \ P \ V \ P \ Jun \ 23th, 2024Introductory Chemical Engineering Thermodynamics ElliottFundamentals Of Chemical Engineering Thermodynamics [EPUB]Fundamentals Of Chemical Engineering Thermodynamics Jan 05, 2021 Posted By ... Chemical Engineering Thermodynamics 3rd Edition. SlerFor GATE 2021 CHEMICAL ENGINEERING For Self-study|IIT Bombay| Books Recommendation For ... Paper F3 Feb 7th, 2021 May 1th, 2024.$

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