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Stanford Research Systems QMS Can Be Found In The Instrument Manual. Mathematical Analysis Of The Quadrupole Mass Filter The X- And Y-motions Of An Ion With Mass M And Charge Q Are Described By The Following Equations Of Motion For The Three Separate Coordinate Axes: $2 \ddot{x} + \frac{2}{r_0^2} x = 0$, $2 \ddot{y} + \frac{2}{r_0^2} y = 0$, $2 \ddot{z} = 0$. $\frac{dx}{dt} = v_x$, $\frac{dy}{dt} = v_y$, $\frac{dz}{dt} = v_z$... 1th, 2024

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Stoichiometry Worksheet #2 (mole-mass, Mass-mole Problems)

Stoichiometry Worksheet #2 (mole-mass, Mass-mole Problems) 1. $N_2 + 2O_2 \rightarrow N_2O_4$

O 4 A. If 15.0g Of N₂O₄ Was Produced, How Many Moles Of O₂ Were Required?
15.0g N₂O₄ 1 Mol N₂O₄ 2 Mol O₂ 92.0g N₂O₄ 1 Mol N₂O₄ = 0.326 Mol O₂
B. If 4.0x10⁻³ Moles Of Oxygen Reacted, How Many Grams Of N₂ Were Needed?
4.0x10⁻³ Mol O₂ 1 Mol N₂ 28 ... 2th, 2024

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SO₃ As A Conversion Factor, We Determine The Mass That This Number Of Moles Of SO₃ Has. The First Step Resembles The Exercises We Did In Section 6.4 "Mole-Mole Relationships In Chemical Reactions". As Usual, We Start With The Quantity We Were Given: 3.59 Mol Fe₂O₃ × 3 Mol SO₃ / 1 Mol Fe₂O₃ = 10.77 Mol SO₃ The Mol Fe₂O₃ Units Cancel, Leaving Mol SO₃ ... 1th, 2024

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CRICKET LAB: OBSERVATIONS ANSWER KEY Name: _____ TABLE 1: Mass Of Crickets (A) Mass Of Each Mass Of Crickets (g) Individual Cricket (g) (A/# Of Crickets)** 21°C (Room Temp) 0.66g 0.11g 30°C (Warm) 0.66g 0.11g 10°C (Cold) 0.66g 0.11g **6 Crickets Were Used In Each Respirometer TABLE 2: Measurement Of Oxygen Consumption In Crickets 1. Temperature Tested (°C): ROOM TEMPERATURE (20°C) Data ... 1th, 2024

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