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Molar Gas Constant $R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$
Molar Volume Of An Ideal Gas ($P = 101.325 \text{ kPa}$, $T = 0 \text{ }^\circ\text{C}$) $V_M = 22.412 \text{ dm}^3 \text{ mol}^{-1}$
Standard Partial Molar Enthalpy $H_B^\circ = 996 \text{ J mol}^{-1}$
Standard Partial Molar En Jan 3th, 2024

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