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Problem Sheet 1 - FREE Chemistry Materials Chemistry: Stoichiometry – Problem

Sheet 1 Directions: Solve Each Of The Following Problems. Show Your Work, Including Proper Units, To Earn Full Credit. 1. Silver And Nitric Acid React According To The Following Balanced Equation: $3 \text{Ag(s)} + 4 \text{HNO}_3 \text{(aq)} \rightarrow 3 \text{AgNO}_3 \text{(aq)} + 2 \text{H}_2\text{O(l)} + \text{NO} \dots$ Feb 1th, 2024.

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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$
 Actual yield = 1.45g Theoretical yield = 5.95g
 $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 \times 16 \text{ g/mol O} = 32.00 \text{ g}$
 $\text{O}_2 \text{ %Yield} = \frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$

34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$
 Percent purity = 62.8%
 pure FeCO_3

Jul 2th, 2024

Practice Work 53 Stoichiometry-04 Mixed Stoichiometry ... Practice Work 53 -

Stoichiometry-04 Mixed Stoichiometry Problems General Information You Will Need A Periodic Table, Your Stoichiometry Notes, And Appendix 12 For This Assignment. Sorry About The Lack Of Format. I'm In A Time Crunch.

123.88 G/mol 70.90 G/mol 137.32 ...

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