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2. What Volume Of 2.50M KCl Is Needed To Make 2.00 L Of A 1.00M Solution?  $M_1V_1 = M_2V_2$   
 $2.50 \cdot 1.00 = 1.00 \cdot V_2$   
 $V_2 = 2.50$   
3. Concentrated H<sub>2</sub>SO<sub>4</sub> Is 18.0M What Volume Is Needed To Make 2.00 L Of 1.00M Solution?  $1.0 \cdot 2.00 = 18.0 \cdot V_2$   
 $V_2 = 0.111$   
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POGIL Activity: Relative Mass And The Mole - 50webs The Mole Is The Number Of Atoms Which Has A Mass In Grams Equal To The Numerical Value Of The Mass Of The Atom In Atomic Mass Units (amu). So The Mass Of An Average Carbon Atom Is 12 Amu And The Mass Of A Mole Of Carbon Atoms Is 12 G. Pogil Answer Key Relative Mass And The Mole Page 1/2 Feb 7th, 2024

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