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Molecular Formula C_2H_4 C_3H_6 C_4H_8 C_6H_{12} C_7H_{14} C_8H_{16} C_9H_{18} $C_{10}H_{20}$
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What Are The Molecularthesis Of The Following Compounds? 1) NaOH 2) H₃PO₄
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Answers: Moles And Stoichiometry Practice Problems Answers: Moles And
Stoichiometry Practice Problems 1) How Many Moles Of Sodium Atoms Correspond
To 1.56×10^{21} Atoms Of Sodium? $1.56 \times 10^{21} \text{ Atoms Na} \times 1 \text{ Mol Na} = 2.59 \times 10^{-3}$
Mol Na $236.022 \times 10 \text{ Atoms Na}$ 2) Determine The Mass In Grams Of Each Of The

Following: A. 1.35 Mol Of Fe $1.35 \text{ Mol Fe} \times 55.845 \text{ G Fe} = 75.4 \text{ G Fe}$ 1 Mol Fe B. 24.5 Mol O Mar 1th, 2024

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