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(b) Divide Molar Mass By Empirical Formula Mass Molar Mass 98.96 G Empirical Formula Mass 49.48 G = 2 = (n) (c) Multiply Empirical Formula By N Obtained Above To Get The Molecular Formula Empirical Formula =  $CH_2Cl$ , N = 2. Hence Molecular Formula Is  $C_2H_4Cl_2$ . Problem 1.3 Calculate The A 2th, 2024

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Question 4. What Do You Think The Last Two Lines Of The Poem Mean? (Looking Back, Does The Poet Regret His Choice Or Accept It?) Answer: The Last Two Lines Of The Poem Talk About Taking A Decision And Accepting Its Consequences- Good Or Bad. The Poet Chose The Road That Was Less Travelled By, Which Indicates He Took A Decision That 1th, 2024

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