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6-(1-bromoethyl)-5-propylnon-4-ene B. 5-(sec-

butyl)-4-(1-chloropropyl)nona-2,6-diene C.

1-chloro-3-ethyl-2-methylhept-2-ene 3th, 2024.

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Ketones And Carboxylic Acids 13. Organic Compounds

Containing Nitrogen 14. Biomolecules 15. Polymers 16.

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 Answers PdfName Each Of The Compounds: (a)  $\text{Ca}$   
 ( $\text{H}_2\text{PO}_4$ ) 2 (b)  $\text{FeSO}_4$  (c)  $\text{CaCO}_3$  (d)  $\text{MgO}$  (e)  $\text{NaNO}_2$  (f)  
 KI Response To A Calcium Dihydrogen Phosphate  
 Answer B Isolated (II) Sulphate Response C Carbonate  
 Football Ansonual Reply Of Magnesium Oxide And  
 Sodium Nitrite Reply F Of Iodide Potassium  $\text{I}_2$ , Contribut  
 1th, 2024Covalent Nomenclature Worksheet  
 AnswersWrite The Formula For Each Covalent  
 Compound. Iodine Trichloride Disulfur Dibromide  
 Arsenic Trioxide Xenon Hexafluoride Write The Formula

For Each Covalent Compound. Boron Trichloride  
Carbon Dioxide Tetraphosphorus Decoxide Germanium  
Dichloride Write Two Covalent 4th, 2024Nomenclature  
Ultimate Review 2017-AnswersPart C — Simple Binary  
Covalent Compounds I. Provide Names For The  
Following Covalent Compounds: 1. NF<sub>3</sub> Nitrogen  
Trifluoride 2. XeO<sub>3</sub> Xenon Trioxide 3. OF<sub>2</sub> Oxygen  
DiFluoride 4. N 4th, 2024.

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2th, 2024Simple Inorganic Formulas And Nomenclature  
Worksheet AnswersFor Example: [FeF<sub>6</sub>]<sup>3-</sup>,  
Hexafluoroferrate (III). Note Here Also That The Latin  
Name For Iron Is Used To Avoid An Awkward Sounding  
Name. A List Of Such Transition Metals Follows In The  
Section On Naming Rules. Ligands Are Ions, Ionic  
Molecules, Or Neutral Molecules That Have Atoms  
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29 4.4 Thermodynamics 30 4.5 Electricity And

Magnetism 31 4.6 Radiation And Light 33 4.7 Acoustics

34 4.8 Quantum Mechanics 34 4.9 Atomic And Nuclear

Physics 35 4.10 Molecular Spectroscopy 37 4.11 Solid

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