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Add Approximately 10 Drops Of Starch Indicator. A Blue-black Colour Should Be Seen As The Starch Reacts With The Residual Iodine. Continue To Add FA 1 1 cm^3 At A Time Until The Blue-black Colour Of The Starch-iodine Complex Disappears And Jan 1th, 2024

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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V^{2+} Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Jan 1th, 2024

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(c) (i) Write A Chemical Equation Representing The
Lattice Energy Of AgBr. (ii) Use The Following Data To
Calculate A Value For The Lattice Energy Of AgBr(s).
First Ionisation Energy Of Silver = $+731\text{kJmol}^{-1}$ Electron
Affinity Of Bromine = 325kJmol^{-1} Enthalpy Change Of
Atomisation Of Silver = $+285\text{kJmol}^{-1}$ Enthalpy Change
Of Atomisation Of Bromine = $+112\text{kJmol}^{-1}$... Jan 1th,
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