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Electrons Or Bonds To Complete Its Octet In The
Neutral State. Hydrogen Has A Valence Of One And
Thus Requires A Single Electron Or Bond To Complete
Its “duet” In The Neutral State. Thus In Hydrocarbons

Carbon Can Form Neutral Bonding Arrangements By
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Character Of Benzene Since, In Reality, All Six Pi
Electrons Are Shared Equally By The Six Carbons. ... It
Should Be Noted That A Number Of Heterocyclic Ring
Systems (conjugated Ring Systems Containing One Or
More Non-carbon Atoms As Members Of The Ring) Are
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Equation (2) The Halogen Is The Larger, More
Polarizable Iodide (better Leaving Group) And This Is
More Readily Displaced By The Nucleophile, But Still Is
A Slow Reaction. In Equation (3) The Halogen Is A Good
Leaving Group (iodide) And It Is Attached To A Benzylic
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