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16. What is the half-life in sec for a first-order reaction with $k = 2.03 \times 10^{-3} \text{ s}^{-1}$? A) 323 B) 369 C) 341 D) 422 E) 395

17. The half-life of a first-order reaction is 45.6 sec. What is the rate constant k in s^{-1} ? A) 0.015 B) 0.014 C) 0.013 D) 0.012 E) 0.011

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