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+ 1 Cl<sub>2</sub> 2 NaCl 4) 2 H<sub>2</sub>O<sub>2</sub> 2 H<sub>2</sub>O + 1 O<sub>2</sub> 5) 1 C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>  
+ 6 O<sub>2</sub> 6 H<sub>2</sub>O + 6 CO<sub>2</sub> 6) 4 H<sub>2</sub>O + 7 CO<sub>2</sub> 1 C<sub>7</sub>H<sub>8</sub> + 9  
O<sub>2</sub> 7) 2 NaClO<sub>3</sub> 2 NaCl + 3 O<sub>2</sub> 8) 4 (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> + 3  
Pb(NO<sub>3</sub>)<sub>4</sub> 1 Pb<sub>3</sub>(PO<sub>4</sub>)<sub>4</sub> + 12 NH<sub>4</sub>NO<sub>3</sub> 4th, 2024

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Fractions As Coefficients In The Chemical Equations?  
Answer: Since There Cannot Be Half Of A Molecule Or  
Half Of An Atom, The Final Set Of Coefficients Has T  
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C. Nuclear Reactions Often Produce Large Amounts Of Energy Because Small Amounts Of Mass Are Converted Into Energy (see Einstein's Famous Equation,  $E=mc^2$ )  
D. All Radioactive Isotopes Decay Completely And Disappear Within A Short Time (1 Year Or Less) 30.  
Sr Predict The Decay Pathway For  $^{90}\text{Sr}$ . (Strontium-88 Is The Most Abundant Stable Isotope For ... 3th, 2024

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The Balancing Force Is 180° Anticlockwise Of This So The Balancing Mass Must Be Placed At An Angle Of 208°. D.J.Dunn 12 Here Is The Solution To Worked Example No.2 Using The Same Method. Mass Radius M R X Mrx  $\theta$  A MA 60 60 MA 0 0 Unknown B 5 75 375 200 75000 90 C 2 50 100 300 30000 30 ... File Size: 173KB Page Count: 12 2th, 2024

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