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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$

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NH₄OH = H₂O + NHA + 3 H₂O + 60 + 026 0026 __ 6 D 20 Al + 02 Al, O,) + 4 DiD
2 DiD + 2 + 2D K₂SO₄ (8) + BaCl₂ + __KCl BaSO₄ CsH₁₂ (D₂ + CO₂ + H₂O Ca
(OH)₂ (N) + NHCl (D²D³/₄D'D¹/₂N D¹ N D ° N N D²D³/₄N) + CaCl NHOHp) VO₅ (4) +
DiD °) DiD ° D D D ° + Na + D¹/₂; ZnI₂ ... Jul 2th, 2024

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