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4.  $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$  5.  $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$

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H<sub>2</sub>SO<sub>4</sub> + BaCl<sub>2</sub> → BaSO<sub>4</sub> + 2HCl  
2H<sub>2</sub>O + 2CO<sub>2</sub> + H<sub>2</sub>O Ca

(OH)<sub>2</sub> (N) + NHCl (D<sup>2</sup>D<sup>3/4</sup>D<sup>1/2</sup>N̄ D<sup>1</sup> N̄ D ° N̄ N̄ D<sup>2</sup>D<sup>3/4</sup>N̄) + CaCl NHOHp) VO<sub>5</sub> (4) +  
DiD °) DiD ° D D D ° + Na + D<sup>1/2</sup>; ZnI<sub>2</sub> ... Jul 2th, 2024

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