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Be Fun To Review The Balancing Chemical Equations. For These Problems, Balance
The Equation First Then Do Stoichiometry As Described In The Question. 7:30 PM - A

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Was Burned In Oxygen And Produced A Mixture Of Carbon Dioxide And Carbon
Monoxide In The Mole Ratio 2.00:1.00. The Volume Of Oxygen Used Was 747.0
Milliliters At 1,092K And 12.00 Atmospheres Pressure. Calculate The Percentage 2th,
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