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X Lab Data . 1. Measure Total # KJ (calorimetry), Using  $Q = C \dots$  Hess's Law . The Total Enthalpy Change For A Chemical Or Physical Change Is The Same Whether It Takes Place In One Or Several Steps. • "state Function" Version 1: A Apr 4th, 2024

## Thermochemistry, Thermodynamics Thermochemistry

Standard States The Thermodynamic Standard State Of A Substance Is Its Most Stable Pure Form Under Standard Pressure (1 Bar) And Some Speci C Temperature (298 K Unless Otherwise Speci Ed). I For A Pure Substance In The Liquid Or Solid Phase, The Standard State Is Th Feb 6th, 2024

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Bonds), But The Answer Would Be The Same Regardless. You Only Need To Account For Bonds That Participate In The Reaction. 9. Bond Energy Per Bond (kJ Per Bond) H-Cl 427 S-Cl 271 H-S 347 Cl-Cl 243  $2 \text{ HCl} + \text{SCl}_2 \rightarrow \text{H}_2\text{S} + 2 \text{ Cl}_2$  Reactants Bond Number Of Bonds/Coefficient Bond Energy (each) Total Energy HCl 2 427 854 SCl<sub>2</sub> 2 271 542 Total 1396 Feb 10th, 2024

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Amount Of Heat Absorbed Or Released During Chemical Or Physical Processes. Jan  
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Jennie Borders 2 Section 17.1 - The Flow Of Energy Energy Is The Capacity To Do  
Work Or Supply Heat. Energy Is The Capacity To Do Work Or Supply Heat. Energy  
Has No Mass Or Volume. Energy Has No Mass Or Volume. Chemical Potential Energy  
Is Energy Stored In Chemicals. Mar 1th, 2024

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Solid ... May 15th, 2021 THERMOCHEMISTRY SECTION 17.2 MEASURING AND  
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Chemical And Physical Processes. ~ ... Feb 14th, 2024

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$C_2H_5OH(l) + 3 O_2(g) \rightarrow 2 CO_2(g) + 3 H_2O(g)$  Cot 31) A 100.0 ml. Sample Of 0.300 M NaOH Is  
Mixed With A 100.0 ml. Sample Of 0.305 M  $HNO_3$  31) In A Coffee Cup Calorimeter. If  
Both Solutions Were Initially At 35.0 °C And The Temperature Of The Resulting  
Solution Was Recorded As 37.0 °C, Determine The  $\Delta H_{rxn}$  (in Units Of kJ/mol). Apr  
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