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$\text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g})$  Cot 31) A 100.0 mL Sample Of 0.300 M NaOH Is Mixed With A 100.0 mL Sample Of 0.305 M  $\text{HNO}_3$  31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 °C And The Temperature Of The Resulting Solution Was Recorded As 37.0 °C, Determine The  $\Delta H_{\text{rxn}}$  (in Units Of kJ/mol). Jan 3th, 2024.

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