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**Introduction To Chemical Engineering: Chemical  
Reaction ...**

1 Chemical Reactions 1.1 Rate Of Reaction And  
Dependence On Temperature We Will Once Again Look  
At The Formation Of Ammonia (NH<sub>3</sub>) From Nitrogen  
And Hydrogen (see Section Chemical Equilibrium Of  
The Thermodynamics Chapter). This Reaction Follows  
The Equation:  $N_2 + 3H_2 \rightleftharpoons 2NH_3$  (1)  $H_0 = 92 \text{ KJ Mol}^{-1}$   
 $S_0 = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$  May 4th, 2024

**Chemical Engineering, CM3510: Chemical  
Reaction ...**

Grading: Exam 1 15% Exam 2 15% Exam 3 15% Final

Exam 20% Research Paper 10% Homework 20% Class Participation 5% Last Year, The Following Grading Scale Was Used:  $A \geq 92$   $A/B \geq 89$   $B \geq 82$   $B/C \geq 80$   $C \geq 71$   $C/D \geq 69$   $D \geq 61$   $F \leq 60$  If Necessary, The Grading Scale May Be Curved Downward. Mar 11th, 2024

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