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3) How Many Distinct P Orbitals Exist In The Second Electron Shell, Where  $N = 2$ ? A) 2 B) 3 C) 4 D) 5 E) 6 Answer: B Diff: 1 Section: 1.2 4) The \_\_\_\_\_ Tells Us That Each Orbital Can Hold A Maximum Of 2 Electrons. A) Aufbau Principle B) Pauli Exclusion Principle C) Hund's Rule Principle D) Jan 7th, 2024

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